

Describing Tables and Graphs

Probably the most important thing to keep in mind when describing graphs and tables is how to convey the information properly and accurately. Surprisingly, relatively few key words or grammar elements are needed when describing trends or movements. When describing graphs, tables, and figures in academic research, it is important to use clear, precise, and objective language. Here are key aspects to consider:

1. General Guidelines

- **Clarity:** Ensure descriptions are easy to understand.
- **Objectivity:** Avoid personal opinions; describe what is evident.
- **Conciseness:** Be brief but comprehensive.
- **Accuracy:** Represent data truthfully without exaggeration or misinterpretation.

2. Common Sentence Structures

Use standard phrases to describe trends, comparisons, and relationships.

Describing Graphs and Trends

- *The graph illustrates/shows/presents...*
- *There is an upward/downward trend in...*
- *The number of X increased/decreased significantly/slightly between [year] and [year].*
- *X remained stable/constant at...*
- *A sharp/gradual/steady increase/decrease was observed in...*
- *The highest/lowest point was recorded in...*
- *X peaked at/reached a high of...*

Describing Tables

- *Table X provides a summary of...*
- *The data in Table X indicates that...*
- *According to Table X, the highest/lowest value is...*
- *A comparison of X and Y in Table X reveals that...*
- *The table demonstrates a clear difference between...*

Describing Figures and Diagrams

- *Figure X depicts/illustrates...*
- *The diagram outlines the process of...*
- *As shown in Figure X, there is a relationship between...*
- *The figure highlights the main components of...*
- *A visual representation of X is provided in Figure X.*

3. Comparing Data

- *Compared to X, Y is significantly higher/lower...*

- While X increased, Y showed a decreasing trend...
- X and Y followed a similar/opposite pattern...
- There was a notable difference between A and B...

4. Summarizing Key Insights

- Overall, the data suggests that...
- In summary, the trend indicates...
- These results highlight a significant correlation between X and Y...
- From the figure, it can be concluded that...

Example 1: Describing a Line Graph

Graph Title: Effect of Temperature on Reaction Rate of Hydrochloric Acid and Magnesium

Description:

"Figure 1 illustrates the effect of temperature on the reaction rate between hydrochloric acid (HCl) and magnesium (Mg). The graph shows that as the temperature increases from 20°C to 60°C, the reaction rate rises significantly. At 20°C, the reaction takes approximately 120 seconds to complete, whereas at 60°C, it is completed within 30 seconds. This trend suggests that higher temperatures lead to increased molecular collisions, thereby accelerating the reaction rate, consistent with the collision theory."

Example 2: Describing a Bar Chart

Graph Title: Solubility of Different Salts in Water at 25°C

Salt	Solubility (g/100 mL of water)
NaCl	36.0
KCl	34.2
CaCl ₂	74.5
MgSO ₄	25.2

Description:

"Table 1 presents the solubility of different salts in water at 25°C. The data indicates that calcium chloride (CaCl₂) has the highest solubility at 74.5 g per 100 mL of water, while magnesium sulfate (MgSO₄) has the lowest at 25.2 g. Sodium chloride (NaCl) and potassium chloride (KCl) exhibit similar solubility levels, at 36.0 g and 34.2 g, respectively. These differences in solubility can be attributed to variations in lattice energy and hydration enthalpy of the salts."

Example 3: Describing a Spectroscopy Graph

Graph Title: UV-Vis Absorption Spectrum of a Dye Solution

Description:

"Figure 2 displays the UV-Vis absorption spectrum of a dye solution. The spectrum shows a strong absorption peak at 520 nm, indicating that the dye primarily absorbs light in the green region and appears red to the human eye. A secondary peak is observed at 300 nm, suggesting additional electronic transitions. The intensity of absorption increases with dye concentration, following Beer-Lambert's law."

Example 4: Describing a Titration Curve

Graph Title: pH Titration Curve of Acetic Acid with NaOH

Description:

"Figure 3 shows the pH titration curve for the reaction between acetic acid (CH_3COOH) and sodium hydroxide (NaOH). Initially, the pH rises gradually as NaOH is added, but a sharp increase occurs near the equivalence point at $\text{pH} \sim 8.7$. Beyond this point, the pH levels off as excess NaOH is introduced. The buffer region is observed between pH 4 and 6, where acetic acid and acetate ions are in equilibrium. This titration curve confirms that acetic acid is a weak acid, neutralized by NaOH to form a slightly basic solution at equivalence."

Example 5: Describing an IR Spectrum

Graph Title: FTIR Spectrum of Aspirin

Description:

"Figure 4 presents the FTIR spectrum of aspirin, displaying characteristic absorption bands. The strong peak at 1750 cm^{-1} corresponds to the $\text{C}=\text{O}$ stretching of the ester functional group, while the broad absorption around 3300 cm^{-1} is attributed to $\text{O}-\text{H}$ stretching. Peaks at 1600 cm^{-1} and 1500 cm^{-1} are associated with aromatic $\text{C}=\text{C}$ stretching. These spectral features confirm the presence of key functional groups in aspirin."

Here are some examples of how to describe graphs, tables, and figures in **materials chemistry** research:

Example 1: Describing an XRD Pattern

Graph Title: X-ray Diffraction (XRD) Pattern of Synthesized ZnO Nanoparticles

Description:

"Figure 1 presents the X-ray diffraction (XRD) pattern of ZnO nanoparticles. The diffraction peaks at $2\theta = 31.7^\circ$, 34.4° , and 36.2° correspond to the (100), (002), and (101) crystal planes, respectively, confirming the hexagonal wurtzite structure (JCPDS No. 36-1451). The sharp and intense peaks indicate a high degree of crystallinity, while the absence of

additional peaks suggests phase purity. Using the Scherrer equation, the average crystallite size was estimated to be 25 nm."

Example 2: Describing a TGA Curve

Graph Title: Thermogravimetric Analysis (TGA) of a Polymer Composite

Description:

"Figure 2 shows the thermogravimetric analysis (TGA) curve of the synthesized polymer composite. The weight loss occurs in three distinct stages: an initial 5% loss below 100°C due to moisture evaporation, a major degradation event between 300–450°C attributed to polymer decomposition, and a final loss above 600°C, indicating char formation. The residual weight of 15% suggests the presence of inorganic fillers, which improve thermal stability."

Example 3: Describing a SEM Image

Figure Title: Scanning Electron Microscopy (SEM) Image of Porous Carbon Material

Description:

"Figure 3 displays the scanning electron microscopy (SEM) image of the synthesized porous carbon material. The image reveals a highly porous morphology with interconnected networks, which can enhance ion transport in electrochemical applications. The pore sizes range from 50 to 200 nm, as estimated by image analysis. The uniform distribution of pores suggests successful templating during synthesis, making this material a potential candidate for supercapacitor electrodes."

Example 4: Describing an Electrochemical Performance Graph

Graph Title: Cyclic Voltammetry (CV) of a NiCo_2O_4 Electrode in 1 M KOH

Description:

"Figure 4 presents the cyclic voltammetry (CV) curves of the NiCo_2O_4 electrode recorded at scan rates of 5, 10, and 20 mV/s in 1 M KOH. The curves exhibit a pair of redox peaks around 0.4 V and 0.55 V (vs. Ag/AgCl), corresponding to the reversible Faradaic reactions of $\text{Ni}^{2+}/\text{Ni}^{3+}$ and $\text{Co}^{2+}/\text{Co}^{3+}$. The peak currents increase with increasing scan rate, indicating a surface-controlled charge storage mechanism. The nearly symmetrical shape suggests good electrochemical reversibility, making NiCo_2O_4 a promising material for energy storage applications."

Example 5: Describing a Mechanical Property Table

Table Title: Mechanical Properties of Polymer Composites

Sample	Tensile Strength (MPa)	Young's Modulus (GPa)	Elongation at Break (%)
Neat Polymer	30.5	1.2	15.8
5% CNT Composite	45.2	2.8	12.3
10% CNT Composite	55.8	4.5	10.1

Description:

"Table 1 summarizes the mechanical properties of polymer composites reinforced with carbon nanotubes (CNTs). The tensile strength increases from 30.5 MPa in the neat polymer to 55.8 MPa for the 10% CNT composite, indicating enhanced load-bearing capacity due to CNT reinforcement. A significant rise in Young's modulus from 1.2 GPa to 4.5 GPa suggests improved stiffness. However, elongation at break decreases with increasing CNT content, likely due to reduced polymer chain mobility. These results highlight the potential of CNTs in enhancing the mechanical performance of polymeric materials."

Example 6: Describing a BET Isotherm

Graph Title: Nitrogen Adsorption-Desorption Isotherm of Mesoporous Silica

Description:

"Figure 5 illustrates the nitrogen adsorption-desorption isotherm of the synthesized mesoporous silica material. The isotherm exhibits a type IV curve with a distinct hysteresis loop in the relative pressure range of 0.4–0.9, characteristic of mesoporous materials. The calculated BET surface area is 850 m²/g, with a pore diameter of ~6.5 nm, determined using the BJH method. These properties indicate a high surface area and well-defined porosity, making this material suitable for catalysis and adsorption applications."

Here are some examples of how to describe graphs, tables, and figures in **materials chemistry** research:

Example 1: Describing an XRD Pattern

Graph Title: X-ray Diffraction (XRD) Pattern of Synthesized ZnO Nanoparticles

Description:

"Figure 1 presents the X-ray diffraction (XRD) pattern of ZnO nanoparticles. The diffraction peaks at $2\theta = 31.7^\circ$, 34.4° , and 36.2° correspond to the (100), (002), and (101) crystal planes, respectively, confirming the hexagonal wurtzite structure (JCPDS No. 36-1451). The sharp and intense peaks indicate a high degree of crystallinity, while the absence of

additional peaks suggests phase purity. Using the Scherrer equation, the average crystallite size was estimated to be 25 nm."

Example 2: Describing a TGA Curve

Graph Title: Thermogravimetric Analysis (TGA) of a Polymer Composite

Description:

"Figure 2 shows the thermogravimetric analysis (TGA) curve of the synthesized polymer composite. The weight loss occurs in three distinct stages: an initial 5% loss below 100°C due to moisture evaporation, a major degradation event between 300–450°C attributed to polymer decomposition, and a final loss above 600°C, indicating char formation. The residual weight of 15% suggests the presence of inorganic fillers, which improve thermal stability."

Example 3: Describing a SEM Image

Figure Title: Scanning Electron Microscopy (SEM) Image of Porous Carbon Material

Description:

"Figure 3 displays the scanning electron microscopy (SEM) image of the synthesized porous carbon material. The image reveals a highly porous morphology with interconnected networks, which can enhance ion transport in electrochemical applications. The pore sizes range from 50 to 200 nm, as estimated by image analysis. The uniform distribution of pores suggests successful templating during synthesis, making this material a potential candidate for supercapacitor electrodes."

Example 4: Describing an Electrochemical Performance Graph

Graph Title: Cyclic Voltammetry (CV) of a NiCo_2O_4 Electrode in 1 M KOH

Description:

"Figure 4 presents the cyclic voltammetry (CV) curves of the NiCo_2O_4 electrode recorded at scan rates of 5, 10, and 20 mV/s in 1 M KOH. The curves exhibit a pair of redox peaks around 0.4 V and 0.55 V (vs. Ag/AgCl), corresponding to the reversible Faradaic reactions of $\text{Ni}^{2+}/\text{Ni}^{3+}$ and $\text{Co}^{2+}/\text{Co}^{3+}$. The peak currents increase with increasing scan rate, indicating a surface-controlled charge storage mechanism. The nearly symmetrical shape suggests good electrochemical reversibility, making NiCo_2O_4 a promising material for energy storage applications."

Example 5: Describing a Mechanical Property Table

Table Title: Mechanical Properties of Polymer Composites

Sample	Tensile Strength (MPa)	Young’s Modulus (GPa)	Elongation at Break (%)
Neat Polymer	30.5	1.2	15.8
5% CNT Composite	45.2	2.8	12.3
10% CNT Composite	55.8	4.5	10.1

Description:

"Table 1 summarizes the mechanical properties of polymer composites reinforced with carbon nanotubes (CNTs). The tensile strength increases from 30.5 MPa in the neat polymer to 55.8 MPa for the 10% CNT composite, indicating enhanced load-bearing capacity due to CNT reinforcement. A significant rise in Young’s modulus from 1.2 GPa to 4.5 GPa suggests improved stiffness. However, elongation at break decreases with increasing CNT content, likely due to reduced polymer chain mobility. These results highlight the potential of CNTs in enhancing the mechanical performance of polymeric materials."

Example 6: Describing a BET Isotherm

Graph Title: Nitrogen Adsorption-Desorption Isotherm of Mesoporous Silica

Description:

"Figure 5 illustrates the nitrogen adsorption-desorption isotherm of the synthesized mesoporous silica material. The isotherm exhibits a type IV curve with a distinct hysteresis loop in the relative pressure range of 0.4–0.9, characteristic of mesoporous materials. The calculated BET surface area is 850 m²/g, with a pore diameter of ~6.5 nm, determined using the BJH method. These properties indicate a high surface area and well-defined porosity, making this material suitable for catalysis and adsorption applications."