

Dr. Sedjelmaci Nesrine  
Lecturer in Toxicology

## METHEMOGLOBIN-FORMING AGENTS

### 1. Definition of MetHb

Methemoglobin (MetHb) is a non-functional hemoglobin in which iron changes from the ferrous state ( $\text{Fe}^{2+}$ ) to the ferric state ( $\text{Fe}^{3+}$ ), which is unable to bind oxygen ( $\text{O}_2$ ).

### 2. Physicochemical properties of MetHb

1. **Spectral properties:** it absorbs in the visible range, which allows its measurement.
2. It is attacked by strong acids and bases (degraded into amino acids).
3. MetHb combines with:
  - $\text{CN}^-$ : cyanmethemoglobin
  - $\text{H}_2\text{S}$ : sulfmethemoglobin
  - $\text{F}^-$ : fluoromethemoglobin
  - $\text{NO}_2$ : nitrosomethemoglobin

### 3. Physiological formation of MetHb

It is continuously formed inside red blood cells. This oxidation is due to normal metabolism (physiological), and the MetHb level normally remains below 2% because it is rapidly reduced by methemoglobin reductase I.

#### Normal levels (physiological):

Adult: 0.5–0.8%

Newborn: 1.5%

Premature: 2%

### 4. Methemoglobin-forming agents

#### *Classification*

#### According to chemical nature

##### 1. Mineral compounds (direct action):

Nitrites ( $\text{NO}_2^-$ ), chlorates,  $\text{KMnO}_4$

## **-Nitrites**

Origin:

- Endogenous: synthesized from the amino acid arginine
- Exogenous: drinking water, animal products, vegetables, fertilizers

Note: nitrites result from the reduction of nitrates by bacteria

Toxic dose:

Maximum allowed concentration in drinking water = 45 mg/L

Toxicity:

Methemoglobinemia, dangerous for young children

## **-Chlorates (NaClO<sub>3</sub> / KClO<sub>3</sub>)**

Origin: herbicide, oral solutions, dyes, explosives, match heads

Toxic dose:

Ingestion of less than 20 matches is not dangerous for a child

Lethal dose: NaClO<sub>3</sub> = 15 g, KClO<sub>3</sub> = 7 g

Toxicity:

Severe methemoglobinemia with hemolysis and anuric tubulopathy

Digestive causticity: nausea, vomiting, epigastric pain

## **-KMnO<sub>4</sub>**

Use: dermatological antiseptic

Toxicity:

Severe methemoglobinemia with hemolysis

Digestive causticity: nausea, vomiting, oropharyngeal pain

## **2. Organic compounds (indirect action):**

Aromatic nitro and amino derivatives, methylene blue (MB: both MetHb-forming agent and antidote), antileprosy drugs (dapson), primaquine, benzocaine, poppers.

### **According to mechanism of action**

- **Direct methemoglobin-forming agents:** high redox potential, act directly on hemoglobin producing superoxide and H<sub>2</sub>O<sub>2</sub>  
Example: ferricyanide/ferrocyanide, hydrazines, nitrites, chlorates

- **Indirect agents:** metabolized into active compounds

Example: nitrobenzene

## 5. Mechanism of action

Fe<sup>2+</sup> forms 6 coordination bonds:

- 4 with tetrapyrrole nitrogen
- 1 with globin histidine
- 1 with O<sub>2</sub>

Fe<sup>3+</sup> forms 6 coordination bonds:

- 4 with tetrapyrrole nitrogen
- 1 with globin histidine
- 1 with H<sub>2</sub>O → oxygen binding becomes impossible

## Consequences of oxidation

- Hemoglobin oxidation → MetHb formation
- Lipid peroxidation by ROS → membrane lysis
- Hemoglobin denaturation → Heinz bodies formation, membrane binding, cell lysis and hypoxia

## Consequences of MetHb in blood

- Brown chocolate coloration of blood
- Free MetHb loses heme → binds albumin → methemalbumin → brown plasma

## 6. Reduction mechanisms of MetHb

- **NADH-dependent pathway (main):** reduces 2/3 of MetHb
- **NADPH-dependent pathway:** requires electron carrier (MB)
- **Glutathione pathway:** secondary (10–15%)
- **Ascorbic acid pathway:** weak, effective if MetHb <15%

## 7. Factors influencing MetHb formation

- Alcohol: increases oxidation, produces ROS
- Fetal hemoglobin: more sensitive
- Digestive disorders: increased sensitivity
- G6PD deficiency: decreased reduction pathways
- Hepatic and renal insufficiency

## 8. Clinical features of poisoning

As MetHb increases → cyanosis

- >10%: cyanosis (bluish nails, lips, mucosa)
- >20%: fatigue, headache, dizziness, tachycardia, polypnea, chocolate-colored blood
- >60%: CNS damage, coma, respiratory depression, cardiac arrest
- >70%: death

## 9. Treatment

**Symptomatic:** oxygen therapy, anticonvulsants

**Decontamination:**

- Skin washing
- Gastric lavage
- Nasal irrigation

**Elimination:**

- Exchange transfusion if severe

## Antidotes

### 1. Methylene blue (MB)

- Mechanism: electron donor → reduces MetHb
- Indication: MetHb > 40%
- Contraindications: G6PD deficiency
- Administration: IV slow
- Side effects: anxiety, dyspnea, hypertension, cyanosis

## **2. Ascorbic acid (Vitamin C)**

- Mechanism: reduction of MetHb
- Less effective but non-toxic
- Used in mild cases

## **10. Toxicological analysis**

Sample: blood (without NaF)

### **Methods:**

- Spectrophotometry (Kaplan method, Evelyn-Malloy)
- CO-oximeter